Product Information:	Product:	Sodium hydride, 60% dispersion in mineral oil, in soluble bags, in resealable cans
	Acros code number:	33214-0000
	CAS number:	7646-69-7
	EINECS number:	231-587-3
	TSCA:	listed
	MDL code number:	MFCD00003471
	Molecular formula:	HNa
	Molecular weight:	23.99 g/mol
Typical Properties:	Appearance:	light grey tacky powder
	Package:	
	Assay:	57 – 63 % NaH
	Melting point of NaH:	425 °C (decomposition)
	Flash point of mineral oil:	165 °C
	Particle size:	5 – 50 µm
	Bulk density:	approx. 0.60 g/cm^3
	Solubility:	reacts with water, and ethanol
	Stability:	stable in dry air up to 230 °C

General Information: Sodium hydride is a very strong base used for condensation reactions like Claisen[1,2,3] and Dieckmann[4,5] condensation, for C, N, O-alkylation, acylation, Aldol addition, synthesis of sodium alcoholates and sodium borohydride, etc.. Since NaH is sensitive to air and humidity, this product has been packaged in bags* which are soluble in common aprotic organic solvents (see below table), and which makes it easy to bring it directly into a chemical reactor without any complicate handling procedure before.

* Material: Poly(styrene-co-butadiene); Thickness: ~60 µm

Solvent	Appearance of the solution at 25 °C	Result
Cyclohexane	clear	suitable
tert-Butyl methyl ether	clear	suitable
Diethyl ether	turbid	suitable
N,N-Dimethylacetamide	turbid	suitable
N,N-Dimethylformamide	turbid	suitable
Ethylene glycol dimethyl ether	clear	suitable
Heptane	bigger parts (turbid at 60 °C)	moderately suitable
Hexane	bigger parts (clear at 60 °C)	moderately suitable
2-Methyltetrahydrofuran	clear	suitable
Tetrahydrofuran	clear	suitable
Toluene	clear	suitable

Please notice that Sodium hydride reacts vigorously with water evolving hydrogen (H_{298} = -132 kJ/mole).

Storage Conditions: Store in a cool dry place (water free area); store in a tightly closed container; keep under a nitrogen blanket.

Material Safety Data Sheet: A **M**aterial **S**afety **D**ata **S**heet (**MSDS**) according to EU guideline 91/155/EWG can be downloaded from our website http://www.acros.com

Literature:

Fieser: **1**, 1075; **2**, 382; **4**, 452; **5**, 610; **6**, 541; **7**, 335; **8**, 458; **9**, 427; **11**, 486; **12**, 447; **14**, 288; **16**, 307.

Merck: **11**, 8573; **12**, 8770; **13**, 8699, **14**, 8625

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2. F.W. Swamer et al.: **Claisen Acylations and Carbethoxylations of Ketones and Esters by Means of Sodium Hydride**, *J. Am. Chem. Soc.*, 1950, 72, 1352-6.
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Note:
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